### **Research Article**

# Design, Synthesis and Antimycobacterial Evaluation Of N- $\frac{\epsilon}{-bromoacetyl}$ norfloxacin.

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#### Abstract

A new derivative of norfloxacin;  $\vee -(\pounds -(\Upsilon - bromoacetyl)piperazin - \Upsilon - yl) - \Upsilon - ethyl - \Upsilon - fluoro - \Upsilon, \pounds dihydro - \pounds - oxoquinoline - \Upsilon - carboxylic acid; was synthesized, characterized by different spectroscopic techniques and evaluated against$ *Mycobacterium tuberculosis*H<sup>T</sup> Rv. The prepared compound exhibited more potent antitubercular activity than norfloxacin.**Key words:**Norfloxacin,*Mycobacterium tuberculosis* 

#### Introduction

Tuberculosis (TB) is considered one of the most chronic bacterial infectious diseases all over the world and the second cause of death due to infectious disease after AIDS.  $^{1,7}$ The most widely used chemotherapy for treatment of uncomplicated TB cases is a combination of isoniazid, pyrazinamide, rifampicin, ethambutol streptomycin. The or emergence multi-drug of resistant tuberculosis (MDR-TB) and extensively resistant tuberculosis(XDRdrug TB)strain<sup>r, i</sup>has added significant difficulty in managing this serious disease. Thus the development of a new potent anti TB drug active against resistance strains has become importance.° of paramount Currently, fluroquinolones moxifloxacin as and ciprofloxacin were approved by WHO as second-line agents in treatment of TB patients and considered one of the main

tools to combat resistant TB strains.<sup>°</sup> Herein, we report synthesis, characterization and evaluation of *N*-acetyl norfloxacin against *Mycobacterium tuberculosis*  $H^{rv}Rv$ .

#### Results and discussion Chemistry

The target compound \ was prepared via Acylation of norfloxacin with bromoacetyl bromide (Scheme 1). The structure of the target compound was confirmed by IR, 'H-NMR. <sup>r</sup>C-NMR and elemental microanalysis. IR spectrum of target compound ' showed appearance of new peak at 1701 related to that of amidic carbonyl(NCO).' H-NMR spectrum showed the appearance of singlet signal at  $\xi$ .  $\gamma$  · ppm , C-NMR showed the appearance of new signal at 7A.79 ppm related to (BrCH<sub>7</sub>) and the elemental micro analysis also, confirm the structure of the target compound.



Scheme : Synthesis of norfloxacin derivative .

## Biology

Results of the anti-TB revealed that the target compound show slightly higher potency against pathogenic *M. tuberculosis* H<sup> $\gamma$ </sup>VRv with MIC; <sup> $\gamma$ </sup>.<sup> $\gamma$ </sup>µM than norfloxacin which has MIC; <sup> $\gamma$ </sup>.<sup> $\gamma$ </sup>µM which might be explained by such increased lipophilicity could enhance the penetration through the lipid rich cell wall of *Mycobacterium*, and lipophilicity is an important consideration in the design and activity of newer antitubecular agents.<sup> $\gamma$ </sup>

#### Experimental Chemistry

Melting points were determined on Stuart electro-thermal melting point apparatus and are uncorrected.IR spectra were recorded on Nicolet iS° FT-IR spectrometer at Minia University.' H NMR spectra were carried out using Bruker apparatus  $\xi \cdot \cdot MHz$ spectrometer, using TMS as internal reference at Sohag University. Elemental analysis was carried out in Al-Azhar University at the regional center for mycology and biotechnology. Reactions were routinely monitored by thin-layer chromatography (TLC) using Merck 9740 pre-coated aluminum plate silica gel (Kieselgel  $(\cdot) \circ x \cdot \cdot$  cm plates with a layer thickness of ... mm, and spots were visualized by exposure to UV-lamp at  $\lambda =$ Yoź nm. Materials: Chemicals and solvents used in the preparation of the target compound are of commercial grade, and purchased from Aldrich, Merck, and El-Nasr pharmaceutical Chemicals Companies.

# Synthesis of the target compound \.

To a stirred solution of norfloxacin (•.٣١٩ g,  $\cdot$ . mmol) in dichloromethane ( $\circ \cdot mL$ ) was added a solution of potassium carbonate ( $\cdot$ .) $\circ$ <sup>7</sup> g,  $\cdot$ .' mmol) in distilled water ( $^{\circ}$ ·ml) at  $^{\circ}$ · $^{\circ}$ C. Then, bromoacetyl (•. ۲۲g, 1.1 bromide mmol) in dichloromethane (Yoml) was slowly added over a period of ". min. Stirring was continued for  $\gamma$  h at  $\cdot$ - $\circ$  °C, then at room temperature for additional  $\gamma\gamma$  h. The whole mixture was then transferred to a separatory funnel where it was extracted with dichloromethane and washed successively with *NHCl* and water. The organic layer was separated, dried over anhydrous sodium sulphate, filtered and the solvent was evaporated under reduced pressure to give compound **\**.

White powder; yield:  $\cdot$ . $\forall \cdot g (\Lambda \xi /); mp$ :  $\gamma_{\xi\gamma}\gamma_{\xi}\gamma_{\xi}$  C: H-NMR ( $\xi \cdot \cdot$  MHz, DMSO $d_{\tau}$ ):  $1.5\circ$  (TH, t, J = V.T Hz,  $NCH_{\tau}CH_{\tau}$ ),  $^{\text{T}.\text{T}A-\text{T}.\text{EY}}$  (<sup>\$</sup>H, m, piperazinyl-H),  $^{\text{T}.\text{V}}$ .  $\forall . \forall \xi$  ( $\xi$ H, m, piperazinyl-H),  $\xi . \forall \cdot$  ( $\forall$ H, s,  $Br\underline{CH}_{\tau}$ ),  $\xi \circ q$  ( $\tau H$ , q,  $J = \nabla \cdot \tau Hz$ ,  $N\underline{CH}$ - $_{\gamma}CH_{\gamma}$ ,  $\forall$ ,  $\forall$ ,  $\forall$ , d,  $J_{H-F} = \forall$ .  $\forall$  Hz, HÅ),  $\forall$ .  $\forall$  $\overline{(1)}$ H, d,  $J_{\text{H-F}} = 1\%$ . Hz, H°),  $\wedge .95$  (1H, s, H<sup> $\gamma$ </sup>), 1°.<sup> $\gamma\gamma$ </sup> (<sup>1</sup>H, s, COOH). <sup> $\gamma$ </sup>C-NMR (1... MHz, DMSO- $d_1$ ):  $1 \leq \sqrt{7}$ ,  $7 \wedge 79$ ,  $\leq 1.71$ ,  $\xi$  1, 1,  $\xi$  9, 01, 1.7, A1, 1.7, 0A, 111, VT (J = YT Hz), 17.1., 177.77, 150.07 (J = 1)Hz),  $1 \leq 9.1 \leq$ ,  $1 \circ 7.7 = 7 \leq A$  Hz),  $17 \circ 29$ , 177.0V and 1V7.7V; Anal. Calcd for  $C_{1A}H_{19}BrFN_{r}O_{\epsilon}$ : C,  $\epsilon_{9.11}$ ; H,  $\epsilon_{.r}o$ ; N, 9.02. Found: C.  $\xi$ 9.72: H.  $\xi$ .71: N. 9.VA.

# Biology

# Anti TB activity

Briefly, the inoculum was prepared from fresh LJ medium re-suspended in VH9-S medium (YH<sup>q</sup> broth, .)% casitone, .o% glycerol, supplemented oleic acid, albumin, dextrose, and catalase [OADC]), adjusted to a McFarland tube No.  $\uparrow$ , and diluted  $\uparrow:\uparrow \cdot;$  $\cdots$  µl was used as inoculum. The compound \ stock solution was thawed and diluted in VH9-S at four-fold the final highest concentration tested. Serial two-fold dilutions of each drug were prepared directly in a sterile 97-well microtiter plate using  $\gamma \cdot \cdot \mu l \forall H^{9}$ -S. A growth control containing no antibiotic and a sterile control were also prepared on each plate. Sterile water was added to all perimetre wells to avoid evaporation during the incubation. The plate was covered, sealed in plastic bags and incubated at  ${}^{\nabla V^{o}}C$  in normal atmosphere. After  $\forall$  days incubation,  $\forall \cdot$  ml of alamar blue solution was added to each well, and the plate was re-incubated overnight. A change in colour from blue (oxidised state) to pink (reduced) indicated the growth of bacteria, and the MIC was defined as the lowest concentration of drug that prevented this change in colour.

# Conclusion

A novel norfloxacin derivative was synthesized, characterized by different spectroscopic and elemental microanalysis techniques and evaluated against  $(H^{\psi} R v)$ .

Result revealed that new compound  $\$  has more potent activity than norfloxacin against ( $H^{\nu}$  Rv).

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# References

- Puratchikody, A.; Natarajan, R.; Jayapal, M.; Doble, M., Chemical biology & drug design Y.YY,VA (1), 9AA-99A; (b) Saraiva, M. F.; De Souza, M. V.; Dau, M. E. T. H.; Araújo, D. P.; De Carvalho, G. S.; De Almeida, M. V., Carbohydrate research Y.Y., Yfo (1), YTY-YTY.
- Y. Lowther, J.; Bryskier, A., Expert opinion on investigational drugs Y...Y, (Y) (Y), YYY-YoA; (b) Velezheva, V.; Brennan, P.; Ivanov, P.; Kornienko, A.; Lyubimov, S.; Kazarian, K.; Nikonenko, B.; Majorov, K.; Apt, A., Bioorganic & medicinal chemistry letters Y.II,YI (Y), 9YA-9Ao.
- Vicente, E.; Pérez-Silanes, S.; Lima, L. M.; Ancizu, S.; Burguete, A.; Solano,

B.; Villar, R.; Aldana, I.; Monge, A., Bioorganic & medicinal chemistry  $\Upsilon \cdot \cdot \P$ ,  $\Upsilon \land \P$ ,  $\Upsilon \land \P$ .

- Corganization, W. H., Treatment of tuberculosis: guidelines. World Health Organization: Υ·۱·; (c) Guerrini, V.; De Rosa, M.; Pasquini, S.; Mugnaini, C.; Brizzi, A.; Cuppone, A. M.; Pozzi, G.; Corelli, F., Tuberculosis Υ·۱٣,٩٣ (٤), ٤·ο-ε11.
- °. Organization. W. Н., Global tuberculosis control: WHO report Y.Y. World Health Organization: Y.Y.; (b) Sotgiu, G.; Migliori, G. B., Pulmonary pharmacology & therapeutics  $1 \cdot 10, TT$ ,  $1 \leq \xi - 1 \leq \lambda$ ; (c) Ng, P. S.; Manjunatha, U. H.; Rao, S. P.; Camacho, L. R.; Ma, N. L.; Herve, M.; Noble, C. G.; Goh, A.; Peukert, S.; Diagana, T. T., European journal of medicinal chemistry 7.10, 1.7, 122-107
- <sup>7</sup>. Rawat, D. S., Medicinal research reviews  $\gamma \cdot \gamma \pi$ ,  $\pi \pi$  ( $\epsilon$ ),  $\gamma \pi \pi \gamma \gamma \epsilon$ .
- V. Foroumadi, A.; Emami, S.; Rajabalian, S.; Badinloo, M.; Mohammadhosseini, N.; Shafiee, A., Biomedicine & Pharmacotherapy Y...9, TY (7), TYT-TY..